

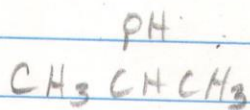
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- #1 a) propan-1-ol
 b) butan-2-ol
 c) cyclobutanol
 d) pentan-2,3-diol
 e) 2,4-heptan-1-ol

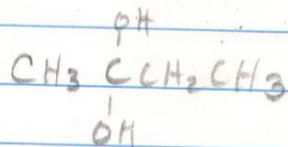
2 a)



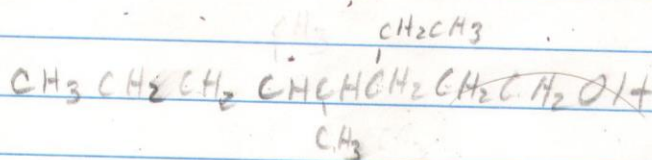
b)



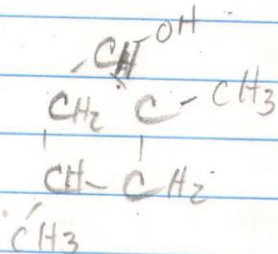
c)



d)



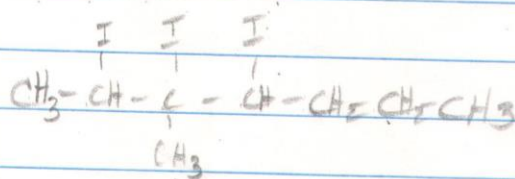
e)



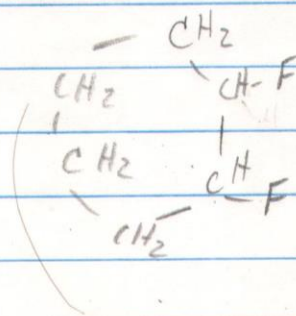
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b)



6 1,2-difluorocyclohexano



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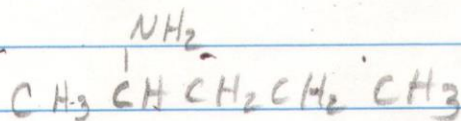
#9 a) methanamine

b) N-ethyl-2,2-dimethylpropanamine

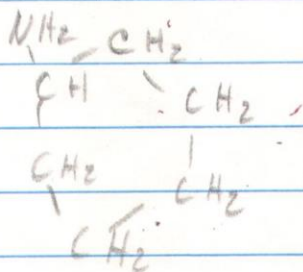
c) butan-2-amine

d) N,N-dimethylcyclopentanamine

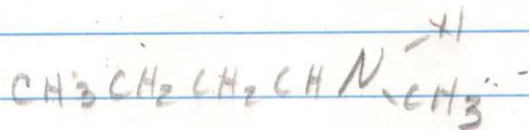
10. a)



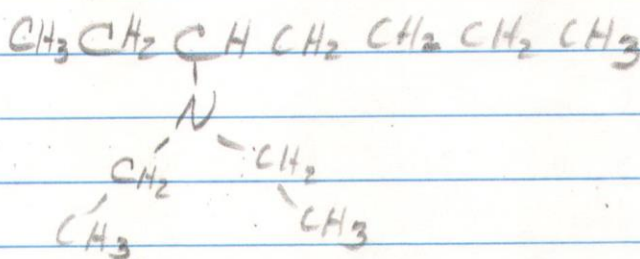
b)



c)



d)



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1a) butan-1-amine

c) 2,3-dimethylcyclopentanol

2. a) heptan-1-ol

b) butan-2-ol

c) ethanamine

d) N-méthylméthanamine

e) 2-chloropropane

3. a) $\text{CH}_3\text{CH}_2\overset{\text{OH}}{\text{CH}}\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$

b) $\text{CH}_3\underset{\text{N}-\text{CH}_2\text{CH}_3}{\text{CH}}\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$

5. a) $\text{H}_3\text{C}-\text{CH}_2-\text{NH}_2 > \text{CH}_3\text{CH}_2\text{OH} > \text{CH}_3\text{CH}_3$

(2 liaisons H)

(1 liaison H)

(force de dispersion)

b) $\text{CH}_3\text{CH}_2\text{CH}_2\text{NH}_2 > \overset{\text{CH}_3}{\text{CH}_3\text{N}-\text{CH}_3} > \text{CH}_3\text{CH}_2\text{CH}_3$

(liaison H)

(dipôle-dipôle)

(force de dispersion)

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#13 a) propanal

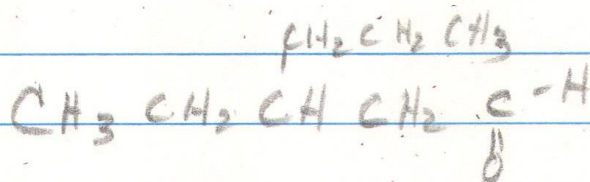
b) octan-3-one

c) 2-methyl hexan-3-one

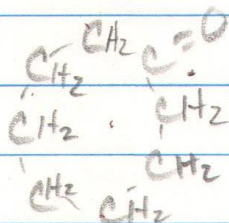
d) 2-ethyl butanal

e) 4-ethyl-3-methyl heptan-2-one

14. a)



b)



c)

